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## Institute of Solid State Physics, Academy of Sciences of the USSR, Chernogolovka Magnetization of Co-H Solid Solutions

## By

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Since a considerable number of hydrogen-containing phases on the basis of transition metals of the groups VI to VIII of the periodic table was obtained only in recent years (see review papers /1, 2/), works dealing with their physical properties are still rare. As to the magnetic properties, it was found that the hydride of antiferromagnetic chromium is paramagnetic down to low temperatures /3/; the hydride of antiferromagnetic manganese is ferromagnetic /4/; the dissolving of hydrogen in ferromagnetic nickel lowers its Curie temperature /5, 6/, and at room temperature nickel hydride is paramagnetic /7/; hydrogenation of strong paramagnetic palladium beyond H to the metal atomic ratio n  $\approx 0.8$  leads to the appearance of superconductivity /8/.

The solubility of hydrogen in the low-temperature h.c.p. allotropic modification of cobalt was shown to increase steadily with pressure of molecular hydrogen, reaching  $n \approx 0.5$  at  $P_{H_2} = 65$  kbar and  $T \approx 500$  K/9/. In the present paper the magnetization of Co-H solid solutions was measured at P = 1 at in a pulsed magnetic field up to 50 kOe in the temperature range (80 to 220)K, the pulse duration being 0.01 s. The specimens were prepared by quenching down to 260 K after the hydrogenation of electrolytically pure cobalt plates  $\approx 0.2$  mm thick at  $P_{H_2} \leq 60$  kbar and T = 520 K for some hours. Note, that at normal pressure Co-H samples thus obtained are kinetically unstable in regard to the desintegration into the metal and molecular hydrogen at temperatures above 260 K/9/. The methods of Co-H sample preparation and their chemical analysis (with an accuracy of about 3%) are described in /9/, and the magnetic measurement techniques in /10, 11/.

The temperature dependence  $\sigma(T)$  of the spontaneous magnetization of our Co and Co-H specimens is weak in the temperature range investigated, and the changings in  $\sigma(T)$  are within the limits of the experimental error  $\delta\sigma/\sigma \approx 0.05$ . The value of



Fig. 1. Spontaneous magnetization  $\sigma'_{0}$  at T = 0 K of Co-H solid solutions as a function of the H to metal atomic ratio n. The broken line shows the  $\sigma'_{0}(n)$  dependence in the rigid band approximation

 $|\partial \sigma / \partial T|$  is known to decrease with decreasing temperature. So, we assumed the mean value of  $\sigma$ 

at 80  $\leq$  T  $\leq$  220 K to be equal to the spontaneous magnetization  $\sigma_0$  at absolute zero.

Experimental results are presented in Fig. 1. The broken line shows the linear dependence  $\sigma_0(n)$  with a slope  $\partial \sigma_0^r / \partial n = -1 \mu_B / \text{Co}$  atom ( $\mu_B$  is the Bohr magneton). This dependence is given by a rigid band approximation under the assumptions that (i) cobalt is a strong itinerant ferromagnet, and (ii) hydrogen, on dissolving into cobalt, transfers its electron to the conduction band of the metal /12/. As one can see from Fig. 1, the slope of the experimental curve  $\sigma_0'(n)$  differs noticeably from the value predicted by the rigid band model even at low hydrogen content in cobalt:  $\partial \sigma_0' / \partial n \Big|_{n=0} \approx 0.6 \mu_B / \text{Co}$  atom. Thus a study of the  $\sigma_0(n)$  dependence shows that pointing out that direct quantum mechanical calculations of the band structures of hydrides of several 3d- and 4d-elements /13 to 15/ also show that the rigid band approximation and, in particular, the "protonic" and "anionic" models are unsatisfactory for describing of physical properties of these compounds.

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