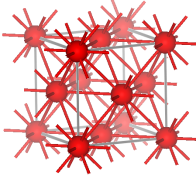
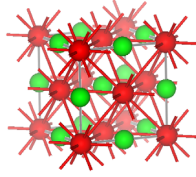
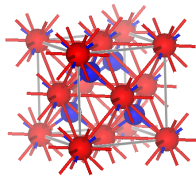


Supplemental Material

for the paper “Solubility of deuterium and hydrogen in fcc iron at high pressures and temperatures” by V. E. Antonov, V. M. Gurev, V. I. Kulakov, M. A. Kuzovnikov, I. A. Sholin, and V. Y. Zuykova

TABLE SI. Compositions and crystal structures used in calculations of the present paper. The relaxed lattice parameters and enthalpies refer to $T = 0$ K and $P = 0$ GPa. In the crystal structures schematically shown in the leftmost column, the metal atoms are red, H(octa) atoms are green, and H(tetra) atoms are blue. The accuracy in calculating the lattice parameters and enthalpies is about 1%. The results in the table are not rounded so that they can be used to calculate ΔV and ΔH for different phases of the same metal, which can be done with higher accuracy due to the mutual compensation of systematic calculation errors.

Structure & space group	Atomic positions (unrelaxed)		Fe-H	Ni-H	Pd-H	Rh-H	
fcc-Me $Fm\bar{3}m$	Me 0 0 0		Lattice parameters (Å)	$a=3.440$	$a=3.533$	$a=3.934$	$a=3.894$
			Volume V (Å ³ /atom Me)	10.17	11.03	15.22	14.77
			Enthalpy H (eV/atom Me)	-865.200	-1354.372	-798.653	-606.861
fcc-MeH(octa) $Fm\bar{3}m$	H(octa) 1/2 1/2 1/2 Me 0 0 0		Lattice parameters (Å)	$a=3.638$	$a=3.745$	$a=4.105$	$a=4.077$
			Volume V (Å ³ /atom Me)	12.04	13.13	17.30	16.94
			Enthalpy H (eV/atom Me)	-881.107	-1370.321	-814.731	-622.743
			$\langle \hbar\omega \rangle_{\text{octa}}^{\text{H}}$ (meV)	114	79	51	65
fcc-MeH(tetra) $F\bar{4}3m$	H(tetra) 1/4 1/4 1/4 Me 0 0 0		Lattice parameters (Å)	$a=3.755$	$a=3.858$	$a=4.218$	$a=4.189$
			Volume V (Å ³ /atom Me)	13.24	14.36	18.76	18.38
			Enthalpy H (eV/atom Me)	-880.765	-1370.300	-814.768	-622.656
			$\langle \hbar\omega \rangle_{\text{tetra}}^{\text{H}}$ (meV)	183	147	124	137

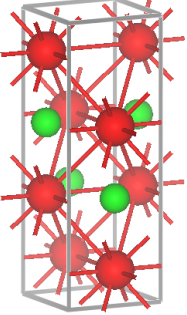
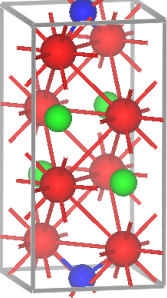
$\text{Me}_2\text{H}(\text{octa})$ $P4/nmm$	$\text{H}(\text{octa})$ 1/2 0 3/8 Me 1/2 0 5/8 Me 0 1/2 7/8		Lattice parameters (Å) $a=2.515$ $a=2.589$ $a=2.828$ $a=2.814$ $c=7.042$ $c=7.241$ $c=8.180$ $c=8.030$	Volume V (Å ³ /atom Me) 11.13 12.13 16.36 15.90
$\text{Me}_8\text{H}(\text{octa})_4\text{H}(\text{tetra})$ $P\bar{4}2m$	$\text{H}(\text{tetra})$ 1/2 1/2 0 $\text{H}(\text{octa})$ 1/4 1/4 5/8 Me 1/4 3/4 1/8 Me 1/4 1/4 3/8		Lattice parameters (Å) $a=3.589$ $a=3.699$ $a=4.047$ $a=4.030$ $c=7.145$ $c=7.321$ $c=8.197$ $c=8.051$	Volume V (Å ³ /atom Me) 11.50 12.52 16.78 16.34

TABLE SII. Calculated hydrogen-induced volume expansions of metal lattice $\beta_{\text{H}}^{\text{O}}$ and $\beta_{\text{H}}^{\text{T}}$ for hydrogen at the O and T interstitial sites, respectively, of the fcc metal lattice. $\beta_{\text{H}}^{\text{O}} = V(\text{fcc-MeH}(\text{octa})) - V(\text{fcc-Me})$ and $\beta_{\text{H}}^{\text{T}} = 8\{V(P\bar{4}2m\text{-Me}_8\text{H}(\text{octa})_4\text{H}(\text{tetra})) - V(P4/nmm\text{-Me}_2\text{H}(\text{octa}))\}$, where V are the atomic volumes from Table SI.

System	$\beta_{\text{H}}^{\text{O}}, \text{Å}^3/\text{atom H}$		$\beta_{\text{H}}^{\text{T}}, \text{Å}^3/\text{atom H}$
	DFT calculation	Experiment	DFT calculation
Fe-H	1.86	2.21 @ 988 K [27]	2.94
Ni-H	2.10	2.20 @ 120 K [1,20]	3.15
Pd-H	2.07	2.47 @ 77 K [1]	3.35
Rh-H	2.18	2.40 @ 120 K [1,20]	3.59

FIG. S1. Calculated phonon dispersions and densities of states for the fcc-MeH(octa) and fcc-MeH(tetra) crystal structures.

