

Supplementary material for: “Synthesis and Characterization of XeAr₂ under High Pressure”

Mengnan Wang¹, Mikhail A. Kuzovnikov¹, Jack Binns², Xiaofeng Li³,

Miriam Peña-Alvarez¹, Andreas Hermann¹, Eugene Gregoryanz^{1,2,4}, Ross T. Howie^{1,2*}

¹*Centre for Science at Extreme Conditions and School of Physics and Astronomy,
University of Edinburgh, Edinburgh, U.K.*

²*Center for High Pressure Science and Technology Advanced Research, Shanghai, China*

³*College of Physics and Electronic Information,
Luoyang Normal University, Luoyang, China*

⁴*Key Laboratory of Materials Physics,
Institute of Solid State Physics, CAS, Hefei, China*

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TABLE S1: Structural parameters of XeAr_2 in the MgZn_2 -type crystal structure at 10 GPa from DFT calculation. Space group $P6_3/mmc$ (No. 194), $a = b = 6.133 \text{ \AA}$, $c = 9.982 \text{ \AA}$.

Atoms	Wyckoff site	x	y	z	Site occupancy	Site Sym
Xe	$4f$	0.33333	0.66667	0.56294	1	$3m$ (100)
Ar1	$2a$	0	0	0	1	$-3m$ (100)
Ar2	$6h$	0.16939	0.33879	0.25000	1	$mm2$ (120)

TABLE S2: Third-order Birch-Murnaghan EoS fittings for XeAr_2 .

Substance	V_0 ($\text{\AA}^3/4.\text{fu}$)	B_0 (GPa)	B_0'	Source
XeAr_2 -Exp	613 ± 55	3.7 ± 1.3	4.9 ± 0.3	This work
XeAr_2 -DFT	600 ± 43	1.6 ± 1.7	5.7 ± 0.7	This work
$4^*\text{Xe} + 8^*\text{Ar}$	672 ± 66	2.8 ± 1.7	5.0 ± 0.4	[1], [2]

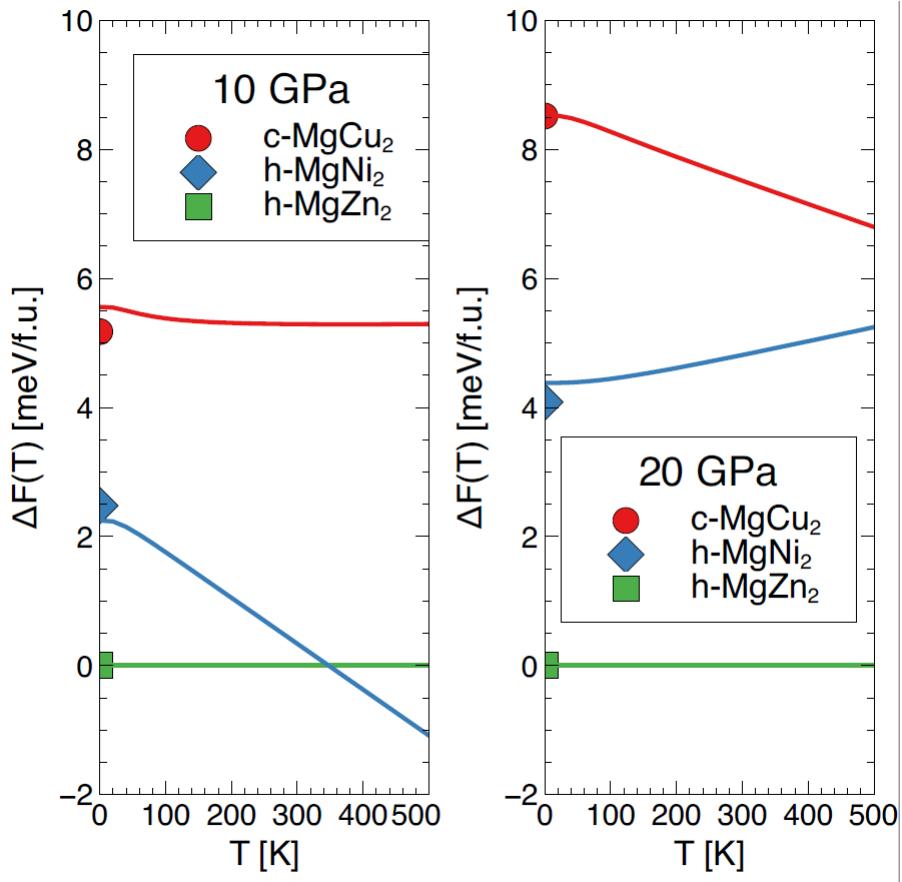


FIG. S1: Relative harmonic vibrational free energies for three structure types of XeAr_2 , plotted against temperature at $P=10$ GPa (left) and 20 GPa (right). Big symbols at $T = 0$ K correspond to relative ground state enthalpies, while lines include zero-point energy (ZPE) and vibrational entropy contributions.

TABLE S3: Optimised volume of XeAr₂ in the MgCu₂ structure type at P = 20 GPa, using various functionals, divided (from top) into parameterisations of LDA, GGA, and dispersion corrections.

Functional	V [Å ³ /f.u.]	V/V _{PBE}	Source
LDA-CA	63.99	0.9274	[3]
LDA-VWN	63.90	0.9261	[4]
PW91	68.79	0.9970	[5]
PBE	69.00	1.0000	[6]
revPBE	71.10	1.0304	[7]
RPBE	71.88	1.0417	[8]
PBEsol	65.88	0.9548	[9]
AM05	67.02	0.9713	[10]
PBE+D2	69.84	1.0122	[11]
PBE+D3	68.76	0.9965	[12]
PBE+D3(BJ)	67.59	0.9796	[13]
PBE+TS	68.85	0.9978	[14]
PBE+TS(HI)	68.76	0.9965	[15]
PBE+MBD	68.43	0.9917	[16]
PBE+dDsC	67.05	0.9717	[17]
PBE+ulg	68.07	0.9865	[18]

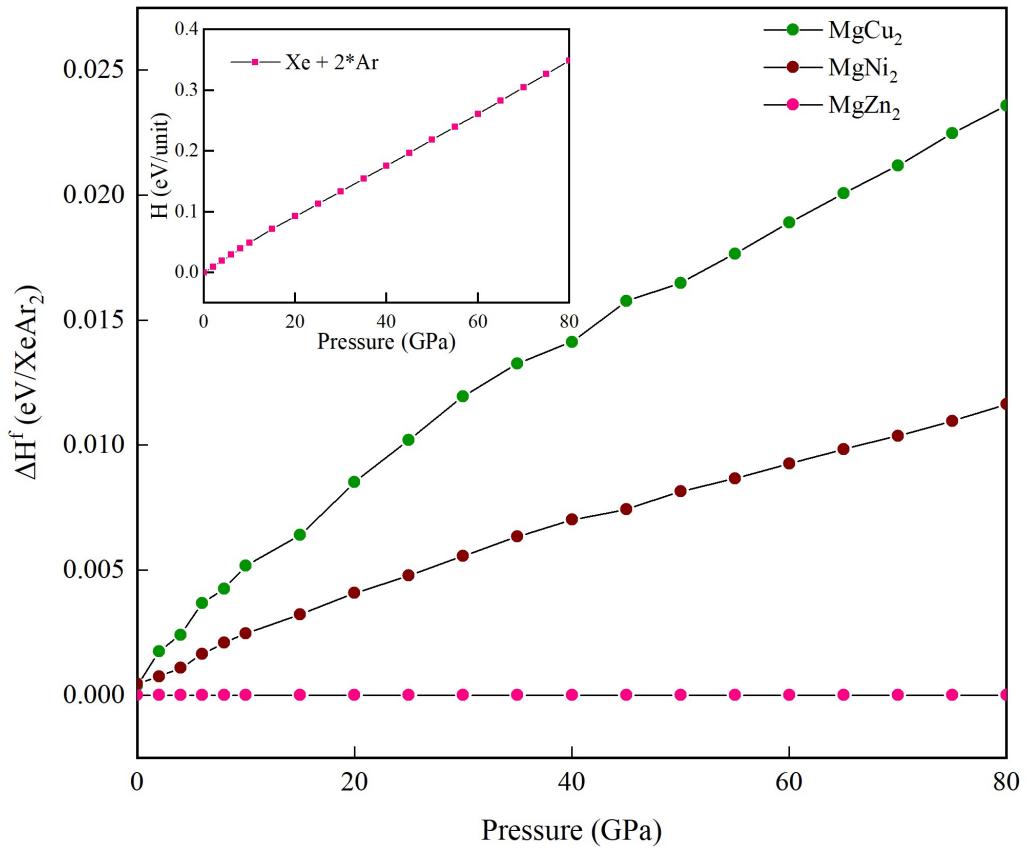


FIG. S2: Ground state formation enthalpies of different XeAr_2 structures, relative to the h-MgZn₂ structure type, as function of pressure, from DFT-PBE calculations. Note the strongly positive relative enthalpy of the constituents $\text{Xe} + 2^*\text{Ar}$.

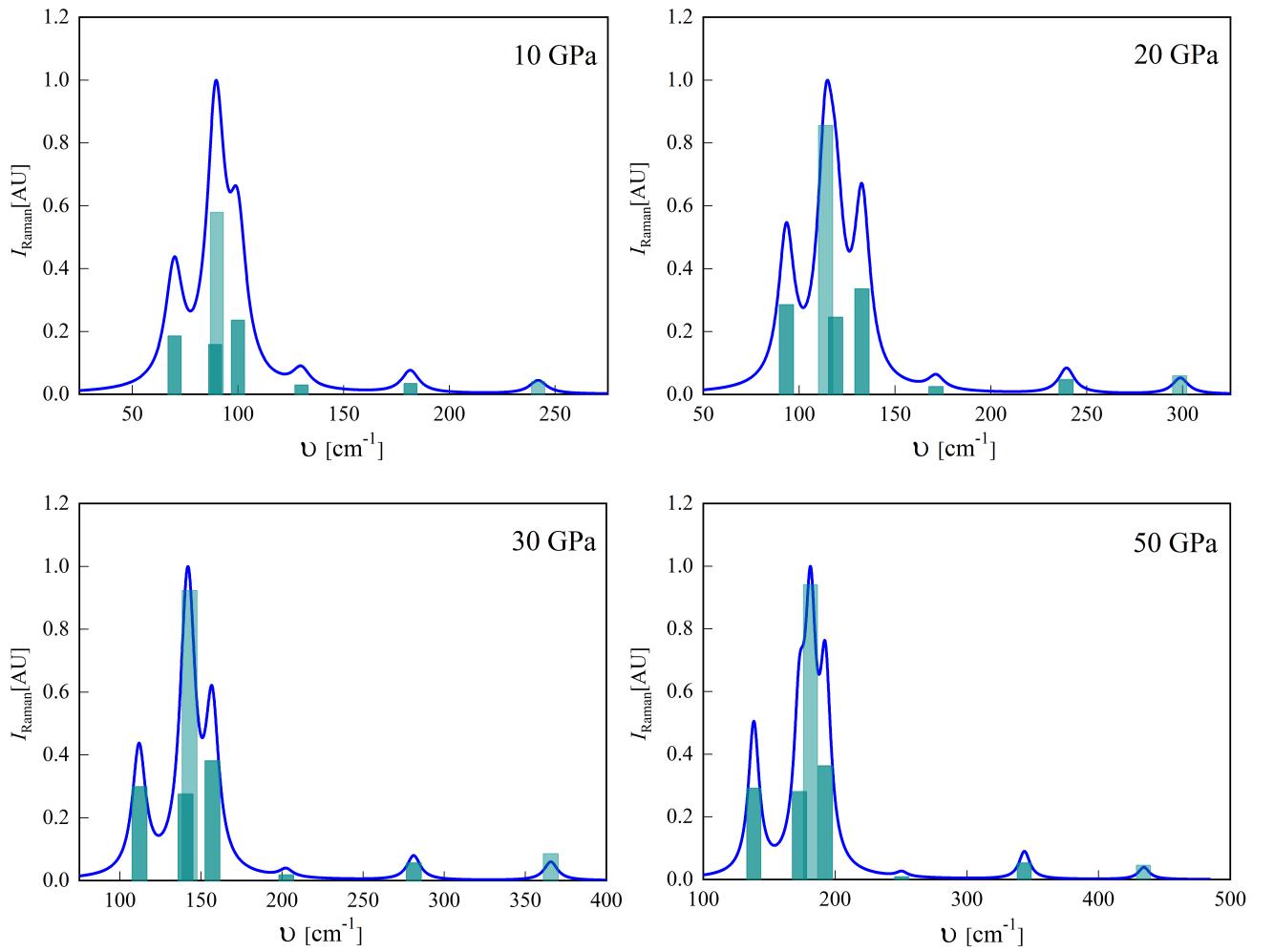


FIG. S3: Simulated Raman spectra of XeAr₂ at 10 GPa, 20 GPa, 30 GPa and 50 GPa, respectively, in the h-MgZn₂ structure type. We have used a fixed FWHM of 10 cm⁻¹ for all modes.

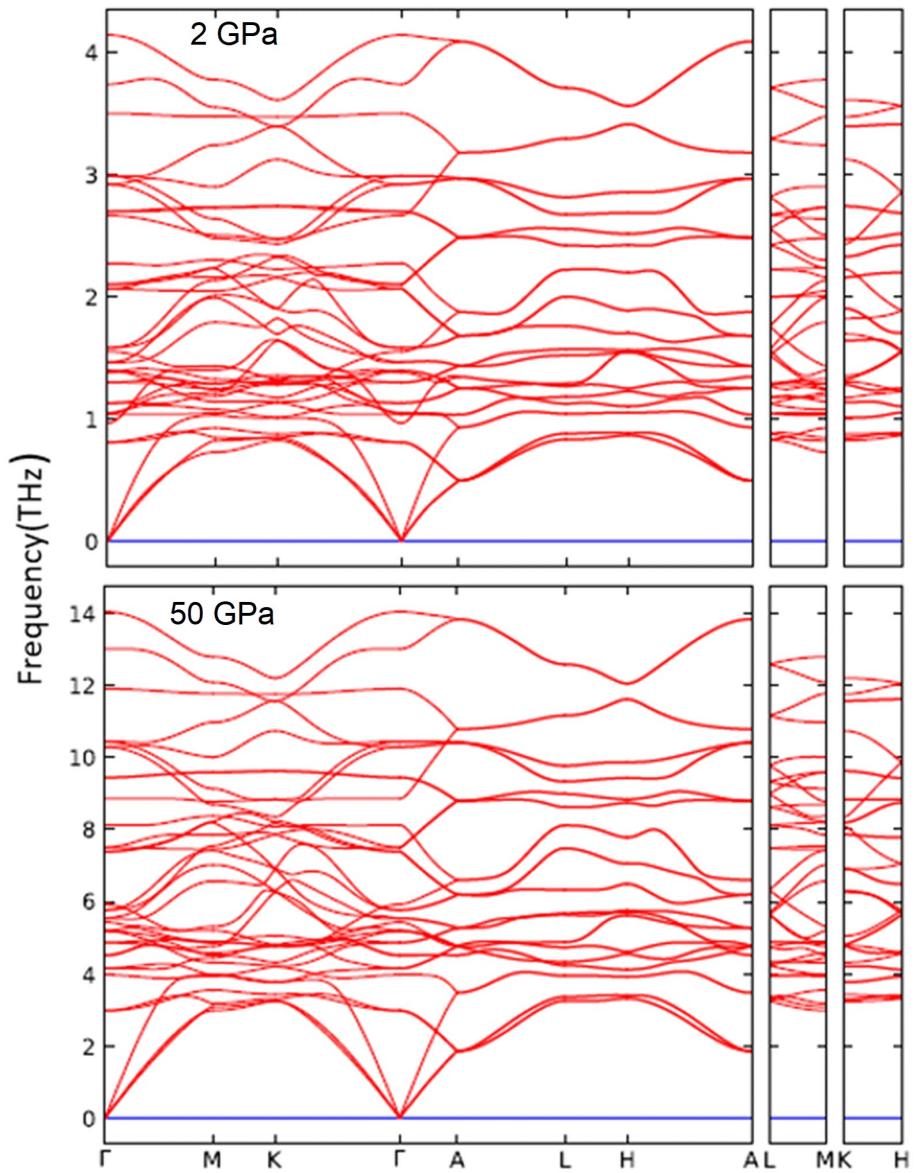


FIG. S4: DFT-PBE phonon dispersions of XeAr_2 at 2 GPa (top) and 50 GPa (lower).

* ross.howie@ed.ac.uk

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